

Module designation	<i>Basic Chemistry for Agriculture</i>
Semester(s) in which the module is taught	<i>1<sup>st</sup></i>
Person responsible for the module	<i>Prof. Dr. Dwi Hapsoro, M.Sc.</i>
Language	<i>Indonesian language</i>
Relation to curriculum	<i>Compulsory</i>
Teaching methods	<i>Lectures (100 minutes) Practicum sessions (170 minutes)</i>
Workload (incl. contact hours, self-study hours)	<i>Contact hours : 14 weeks x 100 minutes Structured learning: 14 weeks x 120 minutes Independent study: 14 weeks x 120 minutes Practicum sessions: 14 weeks x 170 minutes</i>
Credit points	<i>3 (2-1) CP or 4.76 (ECTS) ((14 weeks x 100 minutes) + (14 weeks x 120 minutes) + (14 weeks x 120 minutes) + (14 weeks x 170 minutes)) : 60 minutes/hour = 119 hours : 25 study hours/ECTS = 4.76 (ECTS)</i>
Required and recommended prerequisites for joining the module	-
Module objectives/intended learning outcomes	- <i>Students are able to apply the basic concepts and principles of cultivation technology and the development of sustainable agriculture technology</i>
Content	<i>Basic Chemistry course is a 3 (2-1) credit course. This course presents an understanding of international units, molecules/objects, changes in molecules/objects; properties of atoms/elements in a periodic system, chemical bonds, compounds, and the structure of atoms/elements in molecules/objects; reaction equation equilibrium, solubility, and precipitation reactions; reaction kinetics (speed reactions, catalysts) and molecular movement, laws of thermodynamics; system chemistry of water, acids, bases, salts and buffer solutions; Electrical chemistry (oxidation-reduction); Basic organic chemistry (chain long-chain carbons and functional groups of organic compounds); nitrogen compounds, and lipids.</i>
Examination forms	<i>oral presentation, essay</i>

Study and examination requirements	<p><i>Participants are evaluated based on their performance in class (lectures) (70%) and lab (practicum) (30%).</i></p> <p><i>Performance in theory (100%):</i>  <i>Mid Exam (20%)</i>  <i>Final Exam (20%)</i>  <i>Assignments (40%)</i>  <i>Class participation (10%)</i>  <i>Individual quiz (10%)</i></p> <p><i>Performance in practicum (100%):</i>  <i>Practicum exam (30%)</i>  <i>Pre-test or post-test (10%)</i>  <i>Experiment reports (60%)</i></p>
Reading list	<ol style="list-style-type: none"> <li>1. Chang, Raymond. 2010. Chemistry. 10th ed. McGraw-Hill Companies, Inc., 1221 Avenue of the Americas, New York, NY 10020. 1170p.</li> <li>2. Chang, Raymond. 2010. Chemistry. Student Solution Manual. 10th ed. McGraw-Hill Companies, Inc., 1221 Avenue of the Americas, New York, NY 10020. 698p.</li> <li>3. Petrucci, R.H., F. G. Herring, J.D. Madura, and C. Bissonnette. 2011. General Chemistry: Principles and Modern Applications. Pearson Canada Inc., Toronto, Ontario. 1427p.</li> <li>4. Brown, T.L., H. E. LeMay, B.E. Bursten, C.J. Murphy, Patrick M. Woodward, Matthew W. Stoltzfus. 2018. Chemistry The Central Science. 14th Ed. Pearson Education Limited. Harlow CM17 9NA United Kingdom. 1252p.</li> <li>5. Topping, J.J. K.J. Denniston, and R.E. Caret. 2006. General, Organic, and Biochemistry. Mc Graw Hill Education New York, USA. 87</li> </ol>